Acta Crystallographica Section E

## Structure Reports <br> Online

ISSN 1600-5368

## Mustafa Arslan, ${ }^{\text {a }}$ Erol Asker, ${ }^{\text {b }}$ John Masnovi ${ }^{\text {c }}$ and Ronald J. Baker ${ }^{\text {c }}$

${ }^{\text {a }}$ Department of Chemistry, Faculty of Arts and Sciences, Sakarya University, 54140 Esentepe/ Adapazari, Turkey, ${ }^{\mathbf{b}}$ Necatibey Faculty of Education, Balikesir University, 10100 Balikesir, Turkey, and ${ }^{\text {c }}$ Department of Chemistry, Cleveland State University, OH 44115, USA

Correspondence e-mail: marslan@sakarya.edu.tr

## Key indicators

Single-crystal X-ray study
$T=295 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.006 \AA$
$R$ factor $=0.061$
$w R$ factor $=0.140$
Data-to-parameter ratio $=11.8$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## ( $E, E$ )-1,3-Bis[9,10-dihydro-9-nitro-10-(trinitro-methyl)-9-anthryl]propane

The title compound, $\mathrm{C}_{33} \mathrm{H}_{24} \mathrm{~N}_{8} \mathrm{O}_{16}$, was obtained as a product of the photoreaction between 1,3-di-9-anthrylpropane and tetranitromethane. The molecule occupies a special position on a twofold axis. The trinitromethyl and nitro substituents on the 9,10 -dihydroacridine system are $E$ oriented.

## Comment

Aromatic amine compounds are of interest due to their potential uses as photoconductive materials in a wide range of electrophotographic devices (Hara \& Omae, 1978; Thelakkat, 2002). One common method for the preparation of aromatic amines is the reduction of the corresponding nitro compounds which are commonly prepared through the reaction of aromatic compounds with concentrated nitric acid in the presence of concentrated sulfuric acid. Nitration through the irradiation of the charge-transfer complexes formed between aromatic compounds and tetranitromethane (TNM) offers an alternative route to the use of concentrated acids (Kochi, 1991; Butts et al., 1996; Cox, 1998; Lehnig \& Schürmann, 1998). We have already reported the crystal structure of $(E)-9,10-$ dihydro-9-methyl-9-nitro-10-(trinitromethyl)anthracene as the product of the photoreaction between 9-methylanthracene and TNM (Arslan et al., 2005). In the present paper, we report the crystal structure of the title compound, (I), which is a product of the photoreaction between 1,3-di-9-anthrylpropane, a dimeric analogue to 9 -methylanthracene, with TNM.


The asymmetric unit contains one half-molecule; the other half is generated by a crystallographic twofold axis operation. Bond lengths and angles (Table 1) are similar to those of $(E)$ -9,10-dihydro-9-methyl-9-nitro-10-(trinitromethyl)anthracene (Arslan et al., 2005).

The propylene bridge connecting the two rings shows an anti-anti conformation. The central ring of the 9,10 -dihydroanthracene unit adopts a boat conformation with a dihedral angle between the two benzene ring planes of $25.86(13)^{\circ}$. The trinitromethyl group is attached pseudoaxially at the C10 position of the meso ring. The trinitromethyl and nitro groups on the meso ring are $E$ oriented.

Received 11 July 2006 Accepted 24 July 2006

The crystal packing is mainly determined by van der Waals forces and, contrary to the structure of the monomeric analogue (Arslan et al., 2005), no intermolecular $\pi-\pi$ stacking interactions are observed.

## Experimental

The title compound was synthesized by irradiation for 60 min of a solution of 1,3-di-9-anthrylpropane ( $20 \mathrm{mg}, 0.051 \mathrm{mmol}$ ) and TNM ( $325 \mathrm{mg}, 1.67 \mathrm{mmol}$ ) in a 40 ml pentane $/ 5 \mathrm{ml} \mathrm{CCl}{ }_{4}$ mixture, according to the procedure reported earlier (Arslan et al., 2005). Single crystals suitable for X-ray diffraction studies were grown from a concentrated solution of (I) in chloroform through slow evaporation of solvent at ambient conditions [ $22.36 \%$ yield $(9.0 \mathrm{mg}, 0.0114 \mathrm{mmol})$, m.p. $439-$ $440 \mathrm{~K}]$.

## Crystal data

| $\mathrm{C}_{33} \mathrm{H}_{24} \mathrm{~N}_{8} \mathrm{O}_{16}$ | $Z=4$ |
| :--- | :--- |
| $M_{r}=788.6$ | $D_{x}=1.525 \mathrm{Mg} \mathrm{m}^{-3}$ |
| Monoclinic, $C 2 / c$ | Mo $K \alpha$ radiation |
| $a=23.388(3) \AA \AA$ | $\mu=0.13 \mathrm{~mm}^{-1}$ |
| $b=9.4054(10) \AA$ | $T=295(2) \mathrm{K}$ |
| $c=16.3442(11) \AA$ | Thick plate, colorless |
| $\beta=107.161(7)^{\circ}$ | $0.2 \times 0.2 \times 0.1 \mathrm{~mm}$ |
| $V=3435.2(6) \AA^{3}$ |  |

## Data collection

| Enraf-Nonius CAD-4 | 1481 reflections with $I>2 \sigma(I)$ |
| :--- | :---: |
| $\quad$ diffractometer | $\theta_{\max }=25.1^{\circ}$ |
| $\omega$ scans | 3 standard reflections |
| Absorption correction: none | frequency: 120 min |
| 3049 measured reflections | intensity decay: $2.6 \%$ |

3049 measured reflections
3049 independent reflections

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /[ \sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0444 P)^{2} \\
&+3.4205 P] \\
& \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.21 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.23 \mathrm{e} \AA^{-3}
\end{aligned}
$$

$w R\left(F^{2}\right)=0.140$
$S=1.11$
3049 reflections
258 parameters
H -atom parameters constrained

Table 1
Selected geometric parameters ( $\left(\AA,{ }^{\circ}\right)$.

| $\mathrm{N} 1-\mathrm{C} 13$ | $1.529(5)$ | $\mathrm{N} 9-\mathrm{C} 9$ | $1.561(4)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{N} 2-\mathrm{C} 13$ | $1.534(5)$ | $\mathrm{C} 9-\mathrm{C} 11$ | $1.542(5)$ |
| $\mathrm{N} 3-\mathrm{C} 13$ | $1.538(5)$ | $\mathrm{C} 10-\mathrm{C} 13$ | $1.570(5)$ |
|  |  |  |  |
| $\mathrm{C} 8 A-\mathrm{C} 9-\mathrm{C} 9 A$ | $113.8(3)$ | $\mathrm{C} 9 A-\mathrm{C} 9-\mathrm{C} 11$ | $114.5(3)$ |
| $\mathrm{C} 8 A-\mathrm{C} 9-\mathrm{C} 11$ | $108.8(3)$ | $\mathrm{C} 4 A-\mathrm{C} 10-\mathrm{C} 10 A$ | $113.3(3)$ |

All H atoms were placed geometrically and allowed to ride on their parent atoms with $\mathrm{C}-\mathrm{H}$ distances of $0.93,0.97$ and $0.98 \AA$ for


Figure 1
Molecular structure of (I) with the atom-numbering scheme. Displacement ellipsoids for non-H atoms are drawn at the $30 \%$ probability level. Unlabeled atoms are related to labeled atoms by the symmetry operator (1-x, y, $\frac{1}{2}-z$ ).
aromatic, methylene, and methine H atoms, respectively, and with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$.

Data collection: CAD-4-PC Software (Enraf-Nonius, 1993); cell refinement: CAD-4-PC Software; data reduction: DATRD2 in NRCVAX (Gabe et al., 1989); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and PLATON (Spek, 2003); software used to prepare material for publication: WinGX (Farrugia, 1999).

The authors thank the Turkish Ministry of Education and the CSU College of Graduate Studies for their support of this work.

## References

Arslan, M., Baker, R. J., Masnovi, J. \& Asker, E. (2005). Acta Cryst. E61, o4133-o4135.
Butts, C. P., Eberson, L., Hartshorn, M. P., Robinson, W. T., TimmermanVaughan, D. J. \& Young, D. A. W. (1996). Acta Chem. Scand. 50, 29-47.
Cox, A. (1998). Photochemistry, 29, 164-203.
Enraf-Nonius (1993). CAD-4-PC Software. Version 1.2. Enraf-Nonius, Delft, The Netherlands.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
Gabe, E. J., Le Page, Y., Charland, J.-P., Lee, F. L. \& White, P. S. (1989). J. Appl. Cryst. 22, 384-387.
Hara, S. \& Omae, I. (1978). US Patent 4218247.
Kochi, J. K. (1991). Pure Appl. Chem. 63, 255-264.
Lehnig, M. \& Schürmann, K. (1998). Eur. J. Org. Chem. pp. 913-918.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
Thelakkat, M. (2002). Macromol. Mater. Eng. 287, 442-461.


[^0]:    (C) 2006 International Union of Crystallography All rights reserved

